

2005 Nobel Prize in Chemistry

AWARDED FOR THE DEVELOPMENT OF THE METATHESIS REACTION IN ORGANIC SYNTHESIS

For creative, ingenious and skillful work in the field of olefin metathesis and organometallic chemistry, the Nobel Prize for Chemistry in 2005 was awarded to Yves Chauvin (Institut Français du Pétrole), Robert H. Grubbs (California Institute of Technology) and Richard R. Schrock (Massachusetts Institute of Technology) (1).

The metathesis of alkenes is a remarkable catalytic process in which, under the action of a transition metal compound, denoted by M, two C=C double bonds in two alkene molecules are broken and two new C=C bonds are sequentially formed (2), according to a metallocarbene/metal-lacyclobutane mechanism first formulated by Chauvin (3). The two molecules may be identical (for example, two propene molecules afford ethylene and E/Z-2-butene), or different (for example, $R_2C=CH_2$ and $R'_2C=CH_2$ yield ethylene and $R_2C=CR'_2$). In addition, an intramolecular ring-closing of a diene with two terminal double bonds, accompanied by the release of ethylene, has been developed. The preferred catalysts are based on M = tungsten, rhenium, molybdenum, or ruthenium compounds: with the last two types enjoying widespread applications in both organic and polymer synthesis, mainly due to extensive research by Schrock (4) and Grubbs (5). By varying sub-

stituents on the olefin substrate, a wide range of further olefinic products become readily accessible, while monocyclic or polycyclic olefins provide polymers with unprecedented structures and properties (6).

First observed in the 1950s by H. S. Eleuterio (7), and subsequently by R. L. Banks, G. C. Bailey, W. E. Truett and others, the alkene metathesis reaction (a term coined by Calderon) and the related ring-opening metathesis polymerisation (ROMP) were developed in the following decades by many scientists, each bringing remarkable contributions to this productive area of organic synthesis and catalysis (8).

However, it was only in 1971 that a metal-carbene intermediate was proposed by Y. Chauvin and his student J.-L. Hérisson, to explain – satisfactorily for the first time – the mechanism (3). This extraordinary mechanistic proposal, rationalising Chauvin's astonishing new observations, was immediately embraced by the metathesis community and prompted studies on metal-carbene initiators culminating in the creation of the molybdenum-alkylidene catalysts by R. R. Schrock, see Figure 1 (9), and the 1st and 2nd generation of ruthenium-alkylidene catalysts, by R. H. Grubbs, see Figure 2 (10).

The 2005 Nobel Laureates in Chemistry



Yves Chauvin
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Robert H. Grubbs
Courtesy of Caltech



Richard R. Schrock
photo: L. Barry Hetherington

Some prominent scientists who have worked on metathesis reactions

H. S. Eleuterio, R. L. Banks, G. C. Bailey, W. E. Truett, J. M. Basset, L. Bencze, C. Boelhouwer, M. R. Buchmeiser, N. Calderon, G. Natta, A. F. Noels, Y. Chauvin, A. Demonceau, P. H. Dixneuf, B. A. Dolgoplosk, J. M. Feast, E. Finkelstein, A. Fürstner, R. H. Grubbs, W. A. Herrmann, K. Hummel, H. Hocker, Y. Imamoglu, K. J. Ivin, T. J. Katz, E. Khosravi, B. Marcinek, T. Masuda, J. C. Mol, A. Mortreux, S. P. Nolan, J. J. Rooney, R. R. Schrock, F. Stelzer, R. Streck, E. Thorn-Csanyi, F. Verpoort, K. Wagener, K. Weiss and many others.

These transition metal alkylidene complexes, in themselves a fine artwork in organometallic synthesis, allow extensive practical applications. Some are commercially available and used worldwide, applicable to large-scale or combinatorial syntheses. Full utilisation of these versatile catalysts is just beginning, and by efficient and environmentally-friendly protocols they have opened many ways for obtaining a large variety of organic compounds and natural products with intricate structures, routinely used pharmaceutical intermediates, pesticides, polymers, and composite materials.

Remarkably, the above two types of metathesis catalysts have their own merits and distinctive features thus broadening even more their areas of application. For instance, Schrock's molybdenum-based catalysts, Figure 1, display high activity and selectivity, while Grubbs' ruthenium-based catalysts, Figure 2, are more tolerant to water and functionalities. With these catalysts, the promise of "green chemistry" in organic synthesis is dawning.

In addition to alkene metathesis and ROMP reactions, involving eight electrons ($8e^-$), related processes are now known (11). These formally

involve: $4e^-$ (σ -bond metathesis); $6e^-$ (alkane/alkene insertion); $10e^-$ (alkyne polymerisation, and enyne metathesis) (12), or $12e^-$ (alkyne metathesis).

Acronyms of other reactions based on metathesis are: ADMET (acyclic diene metathesis), RCM (ring-closing metathesis), ARCM (asymmetric ring-closing metathesis) and ROM (ring-opening metathesis).

Ruthenium has featured before in Nobel Prize work (13), and it plays a key role in building modern metathesis catalysts. Reviews reflecting the outpouring of rapid developments in ruthenium-based catalysis for metathesis have been and are being published in this Journal (14).

Lastly, over the past few years, metal-catalysed olefin metathesis has had an enormous impact on organic synthesis, and is emerging as one of the most often used chemical transformations (15). By awarding the 2005 Nobel Prize for Chemistry to Chauvin, Grubbs and Schrock, the outstanding contribution resulting from metathesis for the progress of human society has now been rightfully recognised.

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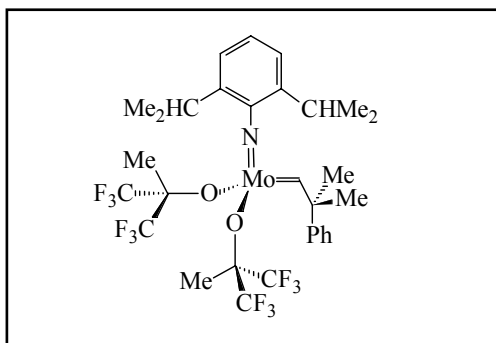


Fig 1. The Schrock catalyst

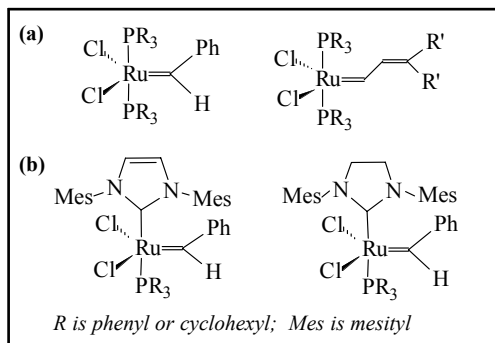


Fig. 2 (a) First, (b) second generation Grubbs' catalysts

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